

Student safety sheets

Mercury and its compounds

including mercury chlorides, sulfides

Substance	Hazard	Comment
Mercury (metal)	TOXIC HEALTH HAZARD	DANGER: fatal if inhaled; causes damage to organs through prolonged or repeated exposure; may damage unborn child. Very toxic to aquatic life. Mercury is very dense and difficult to manipulate, eg in teat pipettes. Containers may be unexpectedly heavy. Elemental mercury vapour is not trapped by filter fume cupboards. Spills need to be cleaned up promptly. The main risk is from inhaling low concentrations of vapour over long periods of time from spills that were not noticed/cleared up. Wear gloves when transferring/dispensing mercury. It forms alloys with gold, silver, etc (eg jewellery), so remove rings.
Mercury(II) chloride Mercury(II) oxide Mercury(I) sulfide Solids Mercury(II) chloride Solution (if 0.1 M or more)	TOXIC CORROSIVE	DANGER: fatal if swallowed or in contact with skin; causes severe burns and eye damage; suspected of causing genetic defects and damaging fertility; causes damage to organs through prolonged or repeated exposure. Very toxic to aquatic life. Older button batteries contain mercury(II) oxide and should be recycled. Mercury compounds used in making Victorian hats caused disease, as in the 'Mad hatter' of Alice in Wonderland.
Mercury(II) chloride Solution (if 0.01 M or more but less than 0.1 M)	TOXIC HEALTH HAZARD	DANGER: fatal if swallowed or in contact with skin; suspected of causing genetic defects and damaging fertility; causes damage to organs through repeated or prolonged exposure. Very toxic to aquatic life.
Mercury(II) chloride Solution (if 0.002 M or more but less than 0.01 M) Mercury(I) chloride Mercury(II) sulfide Solids	HARMFUL ENVIRON. HAZARD	WARNING: harmful if swallowed or in contact with skin; suspected of causing genetic defects and damaging fertility; causes damage to organs through repeated or prolonged exposure. Very toxic to aquatic life. Pollution by mercury compounds in a Japanese river in the 1950s caused serious poisoning of humans who ate river fish (Minimata disease).

Mercury compounds are very toxic. Avoid their use and try to find alternatives.

Typical control measures to reduce risk

- Wear eye protection and suitable gloves; use the lowest possible concentration.
- Avoid the use of mercury compounds where possible (eg, avoid Millon's reagent).
- Transfer/dispense liquid mercury over a tray to contain spills; do not leave mercury surfaces exposed to the air.
- Avoid raising dust (eg, by dampening powder); work in a ducted (not filter) fume cupboard; clear up spills promptly.
- If mercury thermometers are used, take care to avoid breakages. Be prepared in case of spillage.

Assessing the risks

- What are the details of the activity to be undertaken? What are the hazards?
- What is the chance of something going wrong?
- eg solution spurts out of a test tube when heated, mercury metal spills on the floor or a thermometer is broken.How serious would it be if something did go wrong?
- eg could anybody be exposed to dangerous mercury levels for long periods of time?How can the risk(s) be controlled for this activity?
- eg can it be done safely? Does the procedure need to be altered? Should goggles or safety spectacles be worn?

Emergency action

In all emergency situations, alert the responsible adult immediately. Be aware that actions may include the following:

- In the eye
 Irrigate the eye with gently-running tap water for at least 20 minutes. Call 999/111.
 In the mouth/swallowed mouth with drinking water. Do not induce vomiting. Call 999/111.
- Vapour breathed in
 Dangerous only if large amounts of vapour are breathed in over a short period of time (eg when heating metal) or from long-term exposure. Call 999/111.
- Spilt on the skin or clothing
 For mercury metal, remove contaminated clothing and wash off the skin. Check jewellery for damage. For mercury compounds, remove contaminated clothing. Irrigate the affected area with gently-running tap water for at least 20 minutes. Call 999/111 as appropriate. Rinse clothing.
- Spilt on the floor, bench, etc
 For metal, remove jewellery, collect mechanically (eg with syringe). Mop up remainder with a hot paste of 1:1 calcium oxide/sulfur in water. Spread same (dry) mixture over cracks etc. For compounds, scoop up solid. Rinse area with water, diluting greatly. For solutions, use mineral absorbent (eg cat litter).